

Subiectul B. ELEMENTE DE TERMODINAMICĂ

II.a.	$\mu_{\text{CO}_2} = 12 + 2 \cdot 16 = 44 \text{ kg/kmol}$ $N = \frac{m \cdot N_A}{\mu}$ Rezultat final: $N \cong 1,37 \cdot 10^{25}$ molecule
b.	$m_0 = \frac{\mu_{\text{CO}_2}}{N_A}$ Rezultat final: $m_0 \cong 7,31 \cdot 10^{-26}$ kg
c.	$\rho_0 = \frac{p_0 \cdot \mu_{\text{CO}_2}}{RT_0}$ Rezultat final: $\rho_0 = 1,94 \text{ kg/m}^3$
d.	$V_0 = \frac{V_{\mu_0}}{N_A}$ Rezultat final: $V_0 \cong 3,72 \cdot 10^{-26} \text{ m}^3$